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Chapter 4: Atomic Structure Worksheet . Answer the following questions, circle the best answer. 1) Which of the following are isotopes of each other? a) ^{14}C and ^{14}N b) ^3H and ^4He c) ^2H and ^1H d) none of these . 2) The net charge on an atom that has 13 protons, 12 neutrons, and 10 electrons is . a) $5+$ b) -3 c) $3+$ d) -9

Name Chapter 4: Atomic Structure Worksheet Answer the

...

ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4
WORKSHEET PART A Given the following isotopes, determine the

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atomic number, the mass number, the number of protons, electrons and neutrons. Isotope Symbol 131I S31 Uu |He SAl 81W 36Kr SIDU 37Kd Atomic Number Mass Number Protons Electrons Neutrons Isotope Name PARTB.

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Section 4.2 Structure of the Nuclear Atom One change to Dalton's atomic theory is that atoms are divisible into subatomic particles: Electrons, protons, and neutrons are examples of these fundamental particles There are many other types of particles, but we will study these three

Chapter 4 Atomic Structure - Henry County School District

ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4
WORKSHEET PART A Given the following isotopes, determine the atomic number, the mass number, the number of protons, electrons and neutrons. Isotope Symbol ^{131}I ^{238}U ^4He ^{181}W

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36Kr SIDU 37KD Atomic Number Mass Number Protons Electrons
Neutrons Isotope Name PARTB

|He SAI

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Chapter 4 Vocab (12 words) 12 Terms. seversonm. chapter 4
vocab 12 Terms. max220. chapter 4 40 Terms. panzerj140. ...
Science 6.3 worksheet 14 Terms. Mollie_Vick. Bio HUMAN 211
Terms. Mollie_Vick. Science 6.4 Worksheet 18 Terms.
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Chapter Problems Bohr Model Class Work 1. Describe the nuclear
model of the atom. 2. Explain the problems with the nuclear
model of the atom. 3. According to Niels Bohr, what does “n”
stand for? 4. Using wavelengths emitted from a hydrogen atom,

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On: Defining the Atom, Structure of the Nuclear Atom, Distinguishing Among Atoms. Excuse me while I add in my own comments.

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ATOMIC STRUCTURE AND THE PERIODIC TABLE CHAPTER 4
WORKSHEET PART A Given the following isotopes, determine the
atomic number, the mass number, the number of protons,
electrons and neutrons Isotope Symbol ^{131}I ^{31}S ^4He ^{81}W
 ^{36}Kr ^{37}K Atomic Number Mass Number Protons Electrons
Neutrons Isotope

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